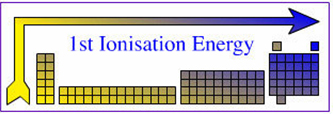
**Describing & Explaining trends in Ionisation Energy (Level 3) examiners tips: Read these please!**

• First ionization energy is the energy required to remove the outermost electron from one mole of an atom in the gaseous state.

• see below a graph summarising the trend in Ionisation Energy



**Also…”don’t be daft”**

you must add in the state of matter as GASEOUS in the chemical equation to represent Ionisation Energy

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